

Long-term impact of elevated CO₂ (FACE) on phosphorus fractions varies in three contrasting soils

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Abstract

Background and aim The long-term effect of elevated CO₂ (eCO₂) on P biogeochemistry in farming systems is largely unknown. An investigation into such an effect is essential for the efficient P management in future climate change. This study compared the effects of eCO₂ on P fractions in three contrasting soils after growing crops for seven years.

Methods A seven-year experiment of free air CO₂ enrichment (FACE) was conducted with a rotation of wheat, field pea and canola grown in intact cores (0.3 m diameter) of Chromosol (Luvisol), Vertosol (Vertisol) and Calcarosol (Calcic Xerosol) under ambient CO₂ (aCO₂) (390 ± 10 ppm) or eCO₂ (550 ± 30 ppm). Crop P removal, changes in soil P fractions and soil biochemical properties were determined.

Results Elevated CO₂ resulted in extra 134, 91 and 93 mg P core⁻¹ removed as grain compared to aCO₂ for Chromosol, Vertosol and Calcarosol, respectively. It decreased the concentration of NaHCO₃-extractable inorganic P in all three soils, and decreased NaOH-extractable inorganic P from 81.8 to 63.1 mg kg⁻¹ in Chromosol, and 22.2 to 5.1 mg kg⁻¹ in Vertosol but did not affect it in Calcarosol. Elevated CO₂ also decreased NaOH-extractable organic P by 20, 12 and 7 mg kg⁻¹ in Chromosol, Vertosol and Calcarosol, respectively. Furthermore, eCO₂ decreased soil organic carbon (by 8.2%) and increased microbial biomass carbon and respiration in the Chromosol but not in other two soils.

Conclusion Long-term eCO₂ favoured microbial mineralization of organic P in the Chromosol and chemical mobilization of non-labile inorganic P in all three soils.

Keywords: C/P ratio; FACE; High atmospheric CO₂; Microbial processes; P fractionation; P removal; SOC

Introduction

Phosphorous (P) cycling in agricultural cropping systems is likely to be impacted by the elevated atmospheric CO₂ concentration (Ochoa-Hueso et al. 2017). Under elevated CO₂ (eCO₂) environments, the demand for P increases in many plant species (Bhattacharyya et al. 2014; Jin et al. 2015). For example, the relative growth response of chickpea (*Cicer arietinum* L.) and field pea (*Pisum sativum* L.) to eCO₂ (550 µL L⁻¹) was more pronounced under P-sufficient than P-deficient conditions (Jin et al. 2012). Moreover, the total P uptake of the plant was greater under eCO₂ compared with ambient CO₂ (aCO₂) although plant P concentration was generally lower due to a dilution effect (Jin et al., 2012; 2013). As grain yields tend to increase and more P contained within the grain exported under eCO₂, the net amount of P in the soil is likely to decrease accordingly in the long term if the soil P is not fully replenished by P fertilizers. Knowledge of the relationship between soil P status and crop removal is essential for development of long-term fertilization strategies to sustain crop productivity in the future eCO₂ environments.

Previous studies have shown that eCO₂ can increase or decrease soil P status. For example, labile soil P increased by approximately 23%, while the recalcitrant P was depleted by 27% after six tree species had been exposed to eCO₂ (700 ppm) for 5 years (Huang et al. 2014). Similarly, compared with aCO₂, 200 ppm higher CO₂ concentration increased soil NaHCO₃-extractable P by 14% when wheat was grown for one season in a loamy soil in a Free Air Carbon Dioxide Enrichment (FACE) system (Zhang et al. 2014). However, a decrease in soil P availability was found after *Deschampsia flexuosa* and *Calluna vulgaris* were exposed to 1 year of eCO₂ in a heathland FACE facility without any P input (Andresen et al. 2010). These inconsistent findings about how eCO₂ affects soil P availability is likely to reflect the various growth responses to eCO₂ among plant species, and different soil physical and biological properties and P fertilizer inputs, all of which influence soil P transformations (Lukac et al. 2010). Little information is available on the long-term impact of eCO₂ on P pools particularly in Australian agricultural soils which are highly weathered and naturally deficient in soil P availability (Lambers et al. 2010; Rossel and Bui 2016).

The dynamics that occur between various P pools in soils is complex. Phosphorus exists in inorganic mineral forms (Pi) and in organic forms (Po) derived from both vegetation and microbial biomass turnover (Turner et al. 2005, Khan et al. 2008). The Pi in soils compose of a continuum of fractions in equilibrium with each other, which differ in their availability to plants. These equilibria are likely regulated by eCO₂-induced changes in chemical processes such as pH. On the other hand, eCO₂ may affect Po fractions through biological immobilization or mineralization due to microbial turnover and release of phosphatase (Richardson and Simpson 2011). Thus, which P fraction eCO₂ may great affect on, to large extent, depends on the accessibility of these P pools by the plant and the effect it has on soil biochemical characteristics. For instance, soil pH in the rhizosphere fundamentally regulates the adsorption-desorption process of Pi (Richardson et al. 2009), and soil microbes interact soil organic matter to mineralize Po (Jin et al. 2015). Therefore, whether chemical or biological processes dominantly contribute to the liberation of non-labile P may vary among soils differing in the composition of P fractions.

This study aimed to investigate the long-term effect of eCO₂ on soil P pools, and associated soil biochemical properties in a wheat-pulse rotation system in three major Australian soils subjected to FACE. We hypothesized that the decrease of P fractions under eCO₂ varied between soils because of differences in both the quantity of P removed as grain harvested and in soil properties in response to eCO₂ would differ between soils, favouring the mobilization

of P from different P fractions to replenish labile P. In particular, Po in soils with high SOC contents (Chromosol) would be mineralized while non-labile Pi would be readily mobilized in soils with low SOC (Vertosol and Calcarosol) in which Po is not a major component in P fractions.

Materials and Methods

Experimental design

A free-air CO₂ enrichment (SoilFACE) experiment was conducted from 2009 to 2015 in a medium-rainfall region, Horsham, Victoria Australia (36°44'57"S, 142°06'50"E). Four FACE bunkers were treated for elevated CO₂ (eCO₂) (550 ± 30 ppm) and the other four bunkers at ambient CO₂ (aCO₂) (390 ± 10 ppm), which represented four replicates. The FACE system used to achieve eCO₂ was detailed in Mollah et al. (2011).

Three major soil types within dryland cropping systems of South-Eastern Australia were involved in this study, i.e. Chromosol, Vertosol and Calcarosol or Luvisol, Vertisol and Calcic Xerosol (FAO-UNESCO 1976). The intact soil cores (0.3 m diameter; 1.0 m depth), which maintain the physicochemical integrity of the soil profile, were collected from paddocks, and secured in PVC sleeves. The intact soil cores were arranged into the bunkers sunk into the ground so that the top of each core was at the soil surface of the surrounding paddock. Each bunker contained up to 44 such cores, twelve of which were dedicated to this study (four for each soil). The soils were sampled at the same time for the analysis of chemical properties, including total carbon (TOC), nitrogen (N), P fractions and pH. The Wimmera region where this experiment was located is characterised by a Mediterranean type climate with cool wet winter and dry hot summers. Over the experimental period, the annual rainfall was in a range from 295-630 mm, maximum mean temperature from 21.4 to 23.1 °C, and minimum mean temperature from 7.2 to 8.1 °C (Figure 1).

A wheat-pulse rotation was deployed in this study. Wheat (*Triticum aestivum* L. cv. Yitpi), field pea (*Pisum sativum* L. cv. PBA Twilight) and canola (*Brassica napus* L. cv. Hyola 50) were grown in the following rotation: field pea in 2009, wheat in 2010, field pea in 2011, wheat in 2012, canola in 2013, wheat in 2014, and canola in 2015. Phosphorus was applied as triple superphosphate at an annual rate of 15 kg P ha⁻¹. Nitrogen in urea was applied at 50 kg N ha⁻¹ for wheat and canola but not for field pea. At maturity, plant shoots were cut-off at the base of the stems. After the removal of grains, the residues were chopped into pieces (less than 2 cm), and returned to their respective mesocosms in the field. Mesh netting was set to avoid the loss of residues during the summer period.

Grain harvest, soil sampling and measurements

Grains/seeds of wheat, field pea and canola were harvested at maturity every year, dried, weighed and stored at 20 °C in plastic containers to avoid moisture absorption. Grain samples were dried at 70°C for 72 h and then ground. Subsamples of ground grain were digested with a mixture of nitric and perchloric acid (4:1) (Yuen and Pollard 1954), and the concentrations of P in digests were analysed using an inductively coupled plasma optical emission spectrometry (ICP-OES) (Perkin Elmer Optima 8000, MA, USA). The concentrations of grain P were then used to calculate total P removal during the experimental period.

In December 2015 following crop harvest, surface soils (0-10 cm) were sampled. A composite sample for each replicate of individual soil types was obtained by combining soil cores from 4 mesocosms of each soil type in each bunker. Around 50 g soil of each sample was subsampled for a pre-incubation at field moisture prior to microbial measurements. Soils

were watered to 60% field capacity, and incubated in plastic bags at 25°C for 15 d. The bags were opened to allow gas exchange with ambient air. The remaining soil was air-dried, and passed through a 0.5-mm sieve for further chemical analyses.

After soil samples were incubated for 15 d, the microbial biomass carbon (MBC) was determined according to Vance et al. (1987). For microbial respiration measurement, 10 g of soil was put into a PVC core (4.5-cm height, 2-cm diameter) with nylon mesh bottom. The PVC core was placed into a sealed mason jar (237 ml) together with a vial containing 2 mL of CO₂-free water to maintain the humidity inside the jar. Soil was incubated for 24 h at 25°C, and then the CO₂ concentration in the air-space of the jar was measured using an infra-red gas analyser (Servomex4210 Industrial Gas Analyser, Cowborough, UK) (Zibilske 1994; Rukshana et al. 2012). The microbial metabolic quotient was calculated as soil respiration rate divided by MBC.

Phosphorus fractionation was performed using air-dried soil using a modified Hedley scheme of P fractionation (Guppy et al. 2000). Inorganic P (Pi) within individual fractions were determined using malachite green (Motomizu et al. 1983). Total P in the bicarbonate (NaHCO₃) and hydroxide (NaOH) fractions were determined following digestion using acid ammonium persulphate in a microwave digester (100 kPa and 120°C) for 45 min (Turner et al. 2003). The Po in these two fractions was calculated by subtracting the Pi from the total P. Total C (SOC) and N concentrations of soils were determined by dry combustion (Perkin Elmer 2400 Series II, USA) following grinding and homogenisation using a ball mill (Retsch MM400, Germany). Soil pH was determined using a pH meter (Thermo Orion 720A+, Beverly, MA, USA) after end-over-end shaking 1 g air-dried soil in 5 ml 0.01 M CaCl₂ for 17 h and centrifugation at 800 g for 5 min.

Statistical analyses

Data were analysed with Genstat 13 (VSN International, Hemel Hempstead, UK). Two-way ANOVA (Steel and Torrie 1980) was performed to determine the effects of CO₂, soil type and their interaction on grain P content, soil pH, SOC, total soil N, microbial properties and soil P fractions. Treatment means were compared using the least significant difference (LSD) at the significant level of $p = 0.05$.

Results

In general, the Chromosol had the highest P total soil P whereas the Calcarosol had the lowest soil P content (Fig. 2). The largest P fraction was NaOH-Po for the Chromosol, residue-P for the Vertosol and NaHCO₃-Pi for the Calcarosol. Compared to the original status in 2009, 7-years of P fertilizer application (by 2015) under aCO₂ significantly ($p = 0.05$) increased the concentrations of all P fractions except for residue-P. Under eCO₂, the concentration of soil P extractable with NaHCO₃ were above the original 2009 level, but decreased all other soil fractions in the Chromosol. In the other two soils, 7-years of P fertilizer application generally increased the concentrations of all the P fractions except for residue-P in three soils, and Po and total P in Calcarosol under eCO₂ (Fig. 2).

Compared to aCO₂, eCO₂ decreased NaHCO₃ extractable Pi in all three soils (Figure 2A). The decrease ranged from 17 to 36% but there was no significant ($p > 0.05$) CO₂ × soil interaction observed (Table 1). The concentration of NaOH-Pi was significantly lower under eCO₂ than that under aCO₂ in the Chromosol and Vertosol soils but not in the Calcarosol (Fig. 2C; Table 1). After 7-years of eCO₂ treatment, NaOH-Pi decreased from 81.8 to 63.1 mg kg⁻¹ in the Chromosol, and from 22.2 to 5.1 mg kg⁻¹ in the Vertosol. Although eCO₂ did

not change NaHCO_3 -extractable Po significantly in the soils (Fig. 2B), the response of NaOH -Po to eCO_2 varied between soils (Fig. 2D). As the largest soil P fraction in the Chromosol, the amount of the decrease in NaOH -extractable Po in this soil reached 20 mg kg^{-1} compared to only 12 and 7 mg kg^{-1} in the Vertosol and Calcarosol, respectively.

Elevated CO_2 decreased HCl-P and residual P with an average decrease of 4.5, 8.2 mg P kg^{-1} across the three soils, respectively (Fig. 2E, 2F). A similar trend was observed in total P in all three soils (Figure 2G, Table 1).

As a total of 742 mg P was applied as fertiliser to each core over the 7 years, the total amounts of P removed by harvested grains under aCO_2 were 382, 462 and 246 mg P per core for the Chromosol, Vertosol, and Calcarosol, respectively (Fig. 3). However, eCO_2 resulted in an extra 134, 91 and $93 \text{ mg P core}^{-1}$ removal compared to aCO_2 for respective soils with a significant $\text{CO}_2 \times \text{soil P}$ interaction ($p < 0.05$) (Table 1).

The concentration of soil organic C (SOC) in the Chromosol decreased significantly ($p < 0.05$) over 7 years, while there was no significant change in the other two soils. In 2015, the concentration of SOC in the Chromosol under aCO_2 was 42 g kg^{-1} , which was 4.8- and 12.1-fold higher than that in Vertosol and Calcarosol, respectively (Fig. 4A). Compared to aCO_2 , eCO_2 decreased SOC concentration by 8.2% in the Chromosol, but did not affect SOC in the other two soils, leading to a significant $\text{CO}_2 \times \text{soil}$ interaction (Table 1). A similar trend was observed in total soil N content (Fig. 4B). However, there was no significant CO_2 effect on C/N ratio in any soil (Fig. 4C). Interestingly, the CO_2 effect on C/P ratio was significant; eCO_2 increased the C/P ratio by 69% in Calcarosol but only by 10% and 12% of increase in Chromosol and Vertosol, respectively (Fig. 4D).

Soil pH varied among the soils with 4.45, 7.32 and 5.75 (extractable in 0.01M CaCl_2) for the Chromosol, Vertosol and Calcarosol soils, respectively. Soil pH did not change significantly over 7 years under aCO_2 . Nevertheless, eCO_2 increased soil pH by 0.2 on average ($P < 0.05$) (Fig. 5; Table 1). The effect of eCO_2 on Microbial Biomass carbon (MBC) depended on the soil type. The MBC in the Chromosol increased from 59 to 75 mg kg^{-1} soil when crops were subjected to eCO_2 , while it was not affected by CO_2 in the Vertosol or Calcarosol soil (Fig. 6A; Table 1). Soil microbial respiration in the Chromosol was 3.8- and 3.3-fold higher than in the Vertosol and Calcarosol soils, respectively (Fig. 6B). Elevated CO_2 increased soil respiration by 24% in the Chromosol and 53% in Calcarosol, but did not affect it in Vertosol. Elevated CO_2 increased the microbial quotient in the Calcarosol but not in the two other soils (Fig. 6C).

Discussion

The amount of P added to the soil:crop system as fertilizers exceeded the P removal in grains over the 7-year study period, contributing to a net accumulation of P in the top 10 cm of soil. The amount of accumulated P however was less than the estimated P remaining in the soil (Figs. 2, 3). This was likely attributed to undecomposed plant residues, and possible P leaching through the soil profile (Sharma et al. 2017), especially in Calcarosol due to low P sorption capacity in sandy topsoil (Siemens et al. 2004; Liu et al. 2012).

When crops were grown under eCO_2 for 7 years, the total P in the topsoil decreased considerably compared to those at aCO_2 (Fig. 2). Greater P removal under eCO_2 contributed to this decrease of total P in soil because larger quantities of grains were harvested under eCO_2 over time (Fig. 3). The increased amount of P removed from the Chromosol was the

largest among the three soils in response to eCO₂, which was consistent with the largest decrease in total P in this soil compared to aCO₂ (Figs. 2, 3). This suggests that crops primarily utilised soil P in this soil layer to produce grain gain under eCO₂. A number of other studies have also shown that eCO₂ increased the growth and biomass of C3 plants and hence P removal when sufficient P was supplied (Jin et al. 2012; Jakobsen et al. 2016). In the present study, although total P decreased under eCO₂ compared to aCO₂, the total P in soils under eCO₂ did not drop significantly ($p > 0.05$) compared to the original soil in 2009. This indicates that the current P fertilizer rate was able to maintain the equilibrium of total P in the soil-plant system.

As P contained in the grain harvested from the cores was the major source of P export, the majority of this P had to pass through the plant-available P pool, i.e. labile P. The eCO₂-induced decrease in P concentration occurred across all soil P fractions measured. Elevated CO₂ decreased the most-readily-available P fraction as well as sparingly-soluble P fractions (Fig. 2). This result was consistent with a previous study showing that eCO₂ decreased the recalcitrant P fractions to replenish the labile-P fraction in a forest ecosystem (Huang et al. 2014). Several studies have reported that the insoluble P associated with Fe and Al oxides and Ca compounds can be depleted by biotic P demand (Chen et al. 2000; Richter et al. 2006). For example, Vu et al. (2008) found that wheat and chickpea grown in a Calcarosol over two growth cycles depleted P in all soil P fractions that were assessed. Thus, it can be deduced from this present study that a long-term exposure to eCO₂ facilitated P uptake by plants from the labile P fractions which in turn accelerated the desorption and/or dissolution of P from the non-labile fractions through biochemical equilibria in soils.

The magnitude of the eCO₂-induced decrease in P fractions, however, depended on soil type. Elevated CO₂ decreased the size of the NaOH-extractable Pi pool in the Chromosol and Vertosol (Fig. 2). The NaOH-Pi is considered as a main fraction representing residual fertilizer P in soils, and P in this fraction is primarily absorbed by sesquioxides (Parfitt 1978; Khan et al. 2008; Zhang et al. 2006), such as Al₂O₃ and Fe₂O₃ in soils (Perrott et al. 1989; Vu et al. 2009). When the consumption of labile P was accelerated by plants grown under eCO₂, NaOH-Pi might be reversibly desorbed into labile Pi to meet increased P demands by crop plants. This view is also supported by previous studies that intensive cropping depleted NaOH-Pi and NaHCO₃-Pi in a number of soils including Vertisol, Mollisols, Ultisols and Oxisols (Guo et al. 2000; Vu et al. 2008). Moreover, the small increase in soil pH under eCO₂ (Fig. 5) in this current study might help to facilitate the desorption of P from the NaOH-Pi fraction in acidic soils, rather than P sorption or precipitation (Gentile et al. 2013). This appeared to occur in the Chromosol where the pH was less than 4.5. However, the size of the NaOH-Pi pool in the Calcarosol did not change greatly in response to eCO₂. This may be attributed to the small size of NaOH-Pi in this particular soil (Fig. 2), limiting the absorption and dissolution of P in this fraction. Furthermore, the majority of P fraction in the Calcarosol was NaHCO₃-Pi, which accounted for more than 40% of total P. The decrease of labile P did not reach the point triggering P transformation from NaOH-Pi because the effectiveness of Al-P/Fe-P in the NaOH-Pi fraction as a P source might depend on the concentration of NaHCO₃-Pi which affected the equilibrium of P between the fractions.

Over 7 years of CO₂ enrichment, the size of the NaOH-Po pool declined in all three soils, especially in the Chromosol in term of the amount of Po decline. This indicates that mineralization of Po became one of major sources of P replenishment of available P, as the NaOH-Po was the primary P fraction in the Chromosol, representing 41% of total P. The Chromosol was more prone to SOC decomposition (Fig. 4), which might be triggered by

increased labile C under eCO₂ (Jin et al. 2015; Vestergard et al. 2016). Since eCO₂ increased plant biomass production in the Chromosol, which had greater SOC concentration than other two soils (Fig. 4), the resultant larger amount of C efflux from the roots in this particular soil might induce a stronger priming effect (van Groenigen et al. 2014; Vestergard et al. 2016). The eCO₂-induced increases in microbial biomass C and microbial activity in the Chromosol (Fig. 6) appears to be associated with a priming effect. Moreover, eCO₂ induced changes in bacterial community composition in plant rhizosphere (Yu et al. 2016), while the change in bacterial community structure and functions (Nie et al. 2013; Pendall et al. 2013) are likely to be associated with decomposition of organic matter (Lian et al. 2017). The biological/biochemical processes during SOC decomposition would lead to P mineralisation (Bünemann et al. 2015). This speculation was supported by the decrease of SOC and NaOH-Po in this soil under eCO₂ (Fig. 4), and the fact that eCO₂ facilitated the transformation from Po to Pi in an Inceptisol (Khan et al. 2008).

Although the NaOH-Po fraction in the Calcarosol was not the major P fraction, this fraction experienced a decrease in size in response to eCO₂ as well (Fig. 2). The mechanism of how microbial activity contributed to the P depletion is most likely different from that of the Chromosol, since the microbial quotient in this soil was greater than that of the Chromosol under eCO₂ compared to aCO₂ (Fig. 6). This indicates that specific microbial groups might be responsible for mineralizing Po. However, the hypothesis is yet to be tested. In the Vertosol, a decrease of NaOH-Po was observed as well, but the biochemical mechanisms are inconclusive due to the insignificant response of microbial properties and SOC under eCO₂ (Figs. 4, 6). As the response of this fraction to eCO₂ was marginal compared to other fractions such as NaOH-Pi and HCl-P, the impact of eCO₂ on P transformation in this soil therefore appears to be mainly associated with physiochemical processes occurring between the P fractions, rather than biological effects.

However, our previous study (Jin et al. 2013) showed that eCO₂ increased NaOH-Po in the rhizosphere of wheat grown in a Vertosol. The discrepancy between this present and previous studies may be attributed to the status of P in response to eCO₂. During the plant growth under eCO₂, the rhizospheric microbes temporarily immobilized P with amount of labile C efflux from roots under eCO₂. The major component of increased Po was likely microbial P associated with diesters such as DNA and/or RNA. Soil microbial biomass has been reported as a labile Po that can contribute to plant available P (Macklon et al. 1997; Kritzler and Johnson 2010). The microbial P would then be re-mobilised over time with the biomass turnover. Strong evidence from both arid ecosystems (Attiwill and Adams 1993) and laboratory microcosms in a grass species (Macklon et al. 1997) suggests that the mineralization of these labile Po can supply large amount of available P to plants. In the long term, net P mineralization may occur following decomposition of the indigenous SOC in soils, such as the Chromosol under eCO₂. Wang et al. (2016) found the significant correlation between Po and C concentrations in Regosols, indicating that mineralizations of Po and organic C were linked each other, and P may be released as the by-product of C mineralization. Thus, the Po compounds bond to soil organic matter such as inositol phosphates may be gradually mineralized due to stimulated phosphomonoesterase with plant succession under eCO₂. However, greater C:P ratios under eCO₂ suggest that eCO₂-induced mineralization of SOC is stronger than their mining Po sources. How the eCO₂-induced priming effect on SOC is associated with P mineralization and how microbial functions contribute to the C/P association warrant further investigation.

Although soil residual P contains occluded inorganic and stable organic forms of P (Chen et al. 2000) and is unlikely to contribute to soil solution P and plant P nutrition in the short-term (McDowell and Condron 2000), long-term exposure to eCO₂ tend to decrease this fraction in three soils. This is consistent with some of the results from Khan et al. (2008) and Huang et al. (2014). The results indicate that greater biotic P demand under eCO₂ leads to a decrease of residual P to sustain soil labile P fractions. Plant growth and P uptake were strongly associated with most P fractions, including the residual P (Vu et al. 2008). A number of studies presented evidence that the depletions in the residual P occurred after exhaustive P uptake (Mckenzie et al. 1992; Zhang et al. 2006). Thus, long-term cropping under eCO₂ likely facilitates their access to recalcitrant P form.

Conclusion

Compared to aCO₂, 7-year exposure of eCO₂ in cropping systems led to a decrease of soil P due to greater crop P removal in three major soils used for grain production in south-eastern Australia. The size and nature of the decrease of P in different fractions varied between the soils. In the Chromosol, the NaOH-Pi and NaOH-Po fractions were the dominant fractions which were decreased by eCO₂, while decreases in NaHCO₃-Pi and NaOH-Pi occurred in the Vertosol. In the Calcarosol, the major decrease of P was recorded in the NaHCO₃-extractable Pi fraction. These results indicate that biochemical processes were involved in the P transformation in the Chromosol while physiochemical processes dominated in the Vertosol and Calcarosol. Due to the significant changes in SOC and Po in the Chromosol, the proper strategies of P management should be particularly concerned in this type of soils to sustain grain production in future eCO₂ environments.

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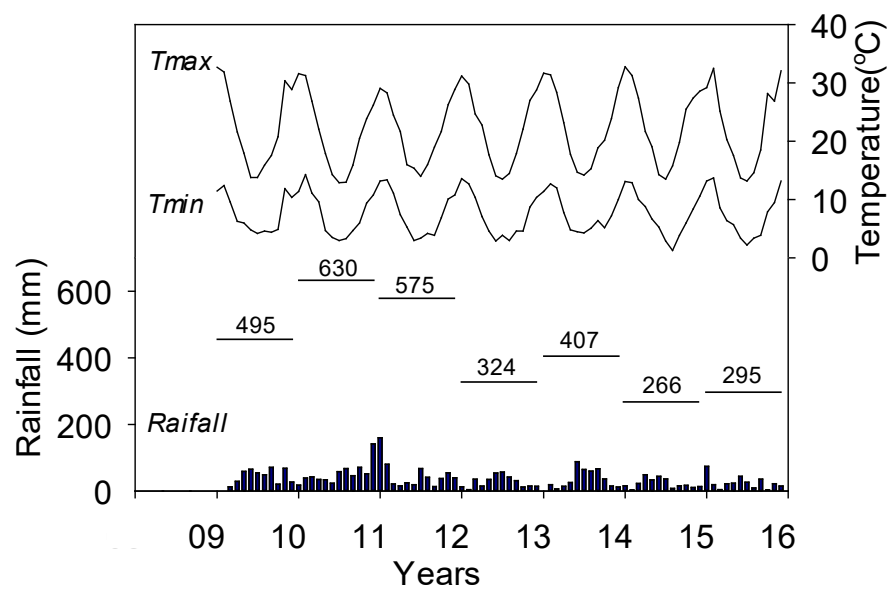


Figure 1. Monthly and yearly rainfalls, and monthly minimal (T_{min}) and maximal (T_{max}) temperatures during the experimental period from 2009 to 2015. The data were obtained from the Bureau of Metreology weather station 10 km from the experimental site.

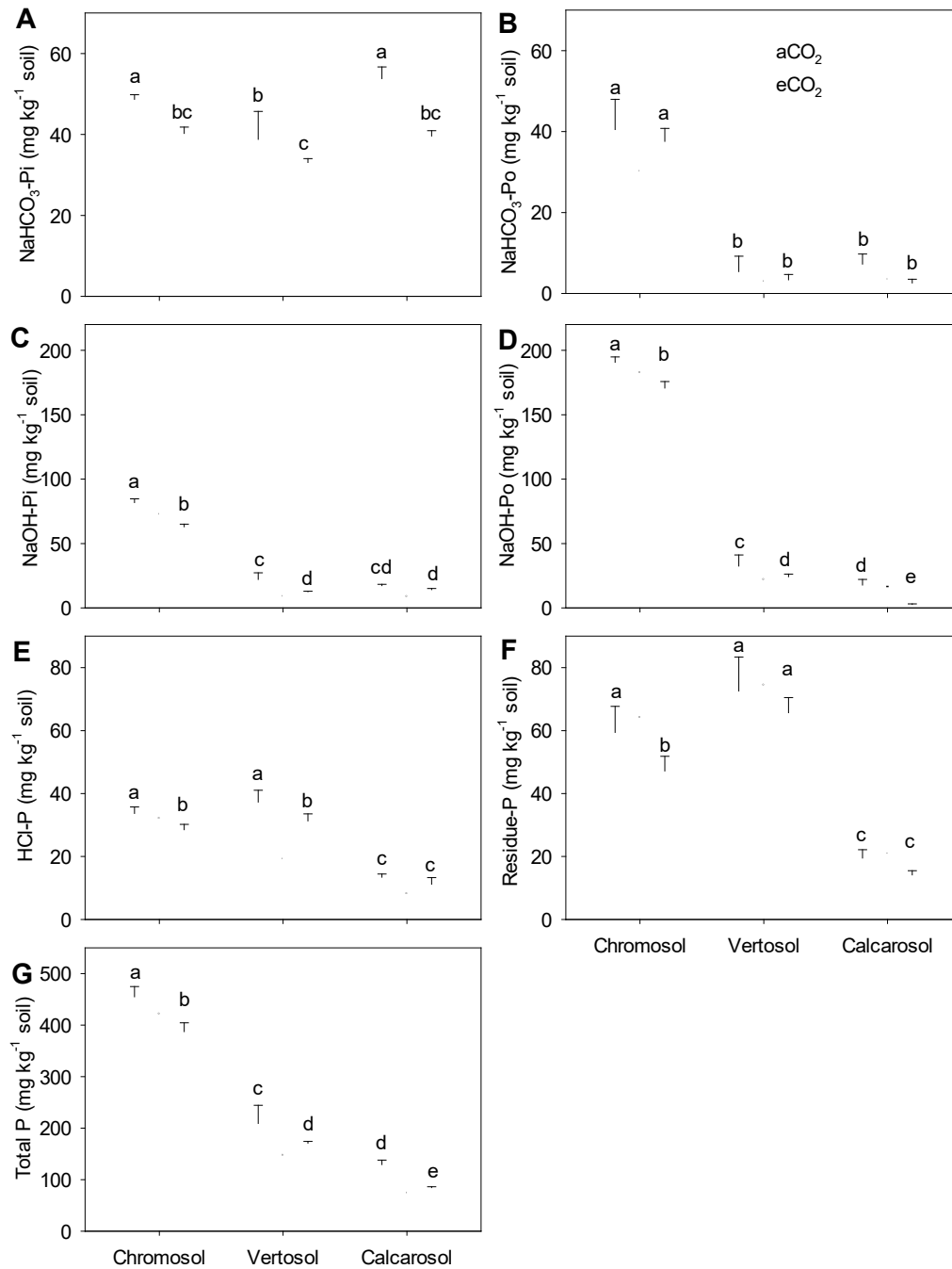


Figure 2. The effect of 7-year eCO_2 on concentrations of $\text{NaHCO}_3\text{-Pi}$ (A), NaOH-Pi (B), $\text{NaHCO}_3\text{-Po}$ (C), NaOH-Po (D), HCl-Pi (E), residual P (F) and total P (G) in the Chromosol, Vertosol and Calcarosol soils. Crops were grown under FACE from 2009 to 2015. Dotted lines represent the initial value in each soil before the start of FACE in 2009. Values were means \pm SE ($n=4$). Values with a same letter are not significantly different between treatments ($p = 0.05$).

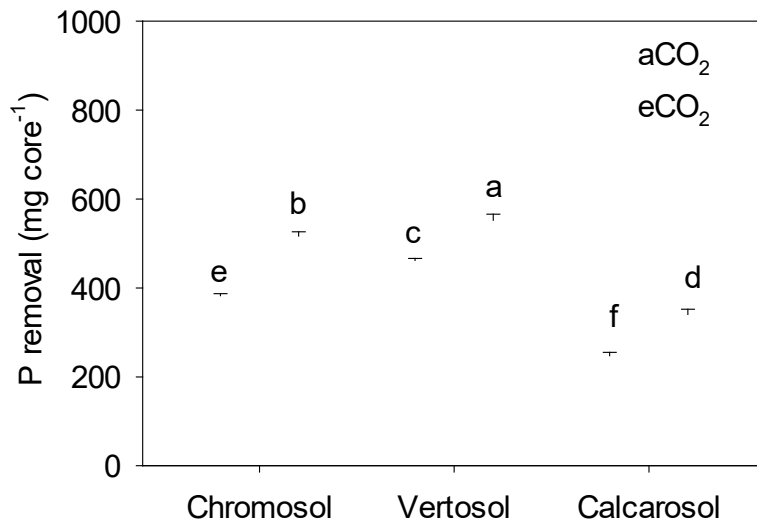


Figure 3. The effect of 7-year eCO₂ on total P removal from the Chromosol, Vertosol and Calcarosol soils over the period from 2009 to 2015 due to the grain harvest. Values were means \pm SE (n=4). Values with a same letter are not significantly different between treatments ($p = 0.05$). Dotted lines represent the total P input as fertilizers in each soil during the experimental period.

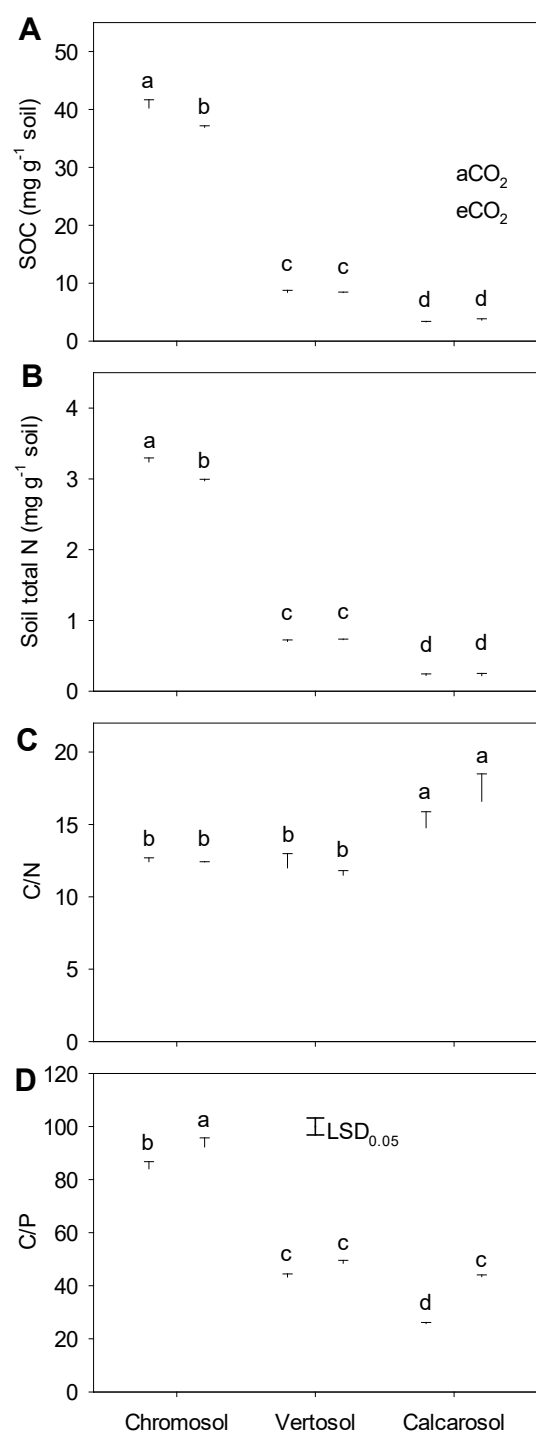


Figure 4. The effect of 7-year eCO₂ on soil organic C (SOC), total soil N, C/N ratio and C/P ratio in Chromosol, Vertosol and Calcarosol. Crops were grown under FACE from 2009 to 2015. Values were means \pm SE (n=4). Values with a same letter are not significantly different between treatments ($p = 0.05$). Dotted lines represent the initial value of each soil before the start of FACE in 2009.

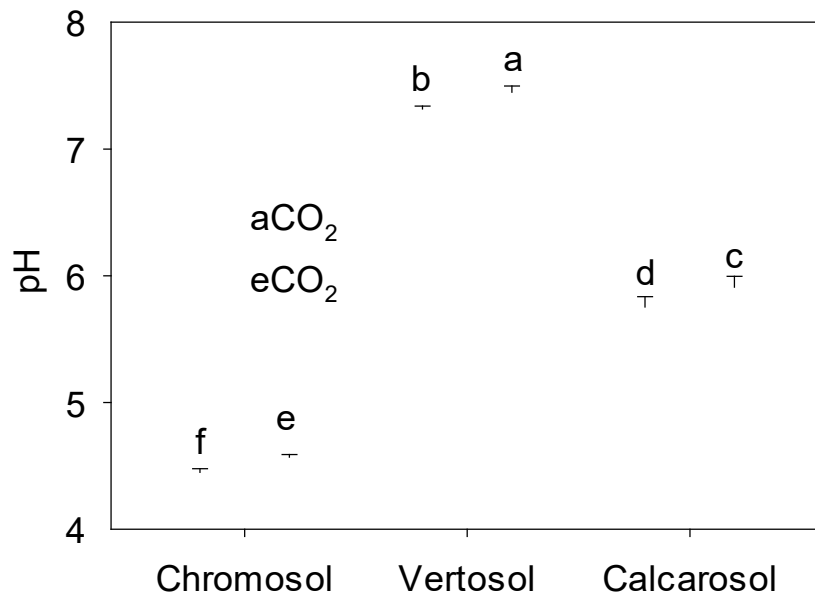


Figure 5. The effect of 7-year eCO₂ on soil pH in Chromosol, Vertosol and Calcarosol. Crops were grown under FACE from 2009 to 2015. Values were means \pm SE (n=4). Values with a same letter are not significantly different between treatments ($p = 0.05$). Dotted lines represent the initial value of each soil before start of FACE in 2009.

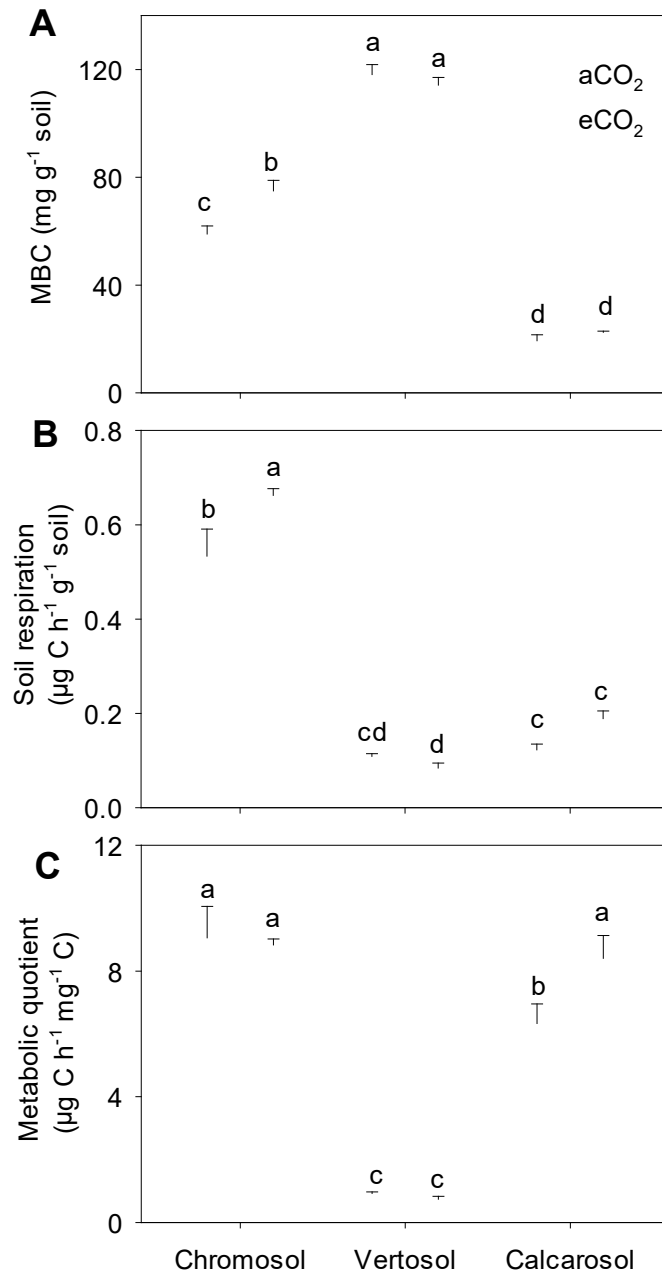


Figure 6. The effect of 7-year eCO₂ on microbial biomass C (MBC) (A), soil basal respiration (B), and metabolic quotient (C) in Chromosol, Vertosol and Calcarosol. Crops were grown under FACE from 2009 to 2015. Values were means \pm SE (n=4). Values with a same letter are not significantly different between treatments ($p = 0.05$).

Table 1. Significant levels of main effects and interactions of CO₂ and soil type on P fractions, P removal, soil organic carbon (SOC), soil N, C/N, C/P, soil pH, and microbial properties.

Variables	CO ₂		Soil		CO ₂ ×Soil	
	M.S.	<i>p</i> values	M.S.	<i>p</i> values	M.S.	<i>p</i> values
NaHCO ₃ -Pi	542	< 0.01	256	< 0.01	39	0.40
NaOH-Pi	661	< 0.001	8321	< 0.001	123	< 0.05
NaHCO ₃ -Po	61	0.31	3150	< 0.001	3.6	0.94
NaOH-Po	808	< 0.001	71718	< 0.001	218	< 0.01
HCl-P	119	< 0.05	1130	< 0.001	7.9	0.67
Residue-P	401	0.12	5766	< 0.001	26	0.84
Total P	15096	< 0.01	221013	< 0.001	830	0.51
P removal	27111	< 0.001	87772	< 0.001	492	< 0.05
SOC	4.6	< 0.05	2160	< 0.001	5.9	< 0.01
Soil N	0.03	< 0.001	14	< 0.001	0.03	< 0.001
C/N	1.1	0.32	29	< 0.001	2.6	0.11
C/P	446	< 0.001	6910	< 0.001	90	< 0.05
pH	0.14	< 0.05	13	< 0.001	0.01	0.49
MBC	304	0.07	15572	< 0.001	325	< 0.05
Soil respiration	0.02	< 0.05	0.60	< 0.001	0.01	< 0.05
Metabolic Quotient	2.9	0.97	141	< 0.001	4.8	< 0.05

M.S., mean square.